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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
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10/734,014

12/11/2003

John G. Nunan

034166.053

2369

25461 7590 10/10/2008  
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EXAMINER

MERKLING, MATTHEW J

ART UNIT

PAPER NUMBER

1795

MAIL DATE

DELIVERY MODE

10/10/2008

PAPER

**Please find below and/or attached an Office communication concerning this application or proceeding.**

The time period for reply, if any, is set in the attached communication.

<b>Office Action Summary</b>	<b>Application No.</b> 10/734,014	<b>Applicant(s)</b> NUNAN, JOHN G.	
	<b>Examiner</b> MATTHEW J. MERKLING	<b>Art Unit</b> 1795	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

### Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

### Status

- 1) ☒ Responsive to communication(s) filed on 04 August 2008.
- 2a) ☐ This action is **FINAL**.                      2b) ☒ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

### Disposition of Claims

- 4) ☒ Claim(s) 1-6,8,9 and 12-26 is/are pending in the application.
- 4a) Of the above claim(s) 14-24 is/are withdrawn from consideration.
- 5) ☐ Claim(s) \_\_\_\_\_ is/are allowed.
- 6) ☒ Claim(s) 1-6,8,9,25 and 26 is/are rejected.
- 7) ☐ Claim(s) \_\_\_\_\_ is/are objected to.
- 8) ☐ Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

### Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on \_\_\_\_\_ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

### Priority under 35 U.S.C. § 119

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All    b) ☐ Some \*    c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
  2. ☐ Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.
  3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

### Attachment(s)

- |  |   |
|--|---|
| 1) <input checked="" type="checkbox"/> Notice of References Cited (PTO-892)          | 4) <input type="checkbox"/> Interview Summary (PTO-413)           |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948) | Paper No(s)/Mail Date. _____                                      |
| 3) <input type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08)          | 5) <input type="checkbox"/> Notice of Informal Patent Application |
| Paper No(s)/Mail Date _____  | 6) <input type="checkbox"/> Other: _____                          |

## **DETAILED ACTION**

### ***Continued Examination Under 37 CFR 1.114***

1. A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on 8/8/08 has been entered.

### ***Claim Rejections - 35 USC § 103***

2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

3. Claims 1-6, 8 and 9 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sung et al. (US 5,981,427) in view of Fujitani et al. (US 4,239,656).

**Regarding claims 1, 2 and 8**, Sung discloses an exhaust treatment device, comprising:

a substrate;

a 1-catalyst layer (col. 10 lines 45-53) deposited on the substrate (col. 8 lines 25-28), the catalyst layer comprising a first catalyst metal (such as Pd, col. 10 lines 33-43) and a

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second catalyst metal (such as Rh, col. 10 lines 33-43), wherein greater than or equal to about 70 wt% of the first catalyst metal and the second catalyst metal is non-alloyed under alloying conditions, wherein the weight percent is based on a combined weight of the first catalyst metal and the second catalyst metal (see col. 8 lines 36-43 which discloses a most preferable embodiment is where greater than 75% of the first and second noble metal components are separate in the later, i.e. will not alloy); and

wherein the first catalyst metal and the second catalyst metal are different and individually selected from the group consisting of platinum, palladium, rhodium, iridium, rhenium, ruthenium, and osmium (col. 10 lines 33-43),

wherein the catalyst layer further comprises an aluminum oxide (col. 7 lines 39-47 and an oxygen storage component (see col. 14 lines 6-12),

wherein the oxygen storage component is represented by the formula  $(\text{Ce}_a\text{Zr}_b\text{La}_c\text{Y}_d\text{Pr}_e\text{O}_x)$ , wherein subscripts a, b, c, d, e, and x, represent atomic fractions, and wherein  $a+b+c+d+e=1$  (see col. 14 lines 6-12, which discloses a composition of oxygen storage material that reads on the claimed composition).

Sung, however, fails to explicitly disclose the aluminum oxide and the storage component have average pore diameters of about 150Å to about 1,000Å, and

Fujitani also discloses a catalyst for purifying exhaust gases and a carrier for the catalyst (see title).

Fujitani, similar to Sung, teaches a catalyst support ( $\gamma\text{-Al}_2\text{O}_3$ , see Example 3 or Example 5 of Fujitani) with an oxygen storage component, also similar to Sung (cerium, col. see Example 3 or Example 5 of Fujitani) and an average pore diameter of 400Å (0.04

μm, see Example 3 or Example 5 of Fujitani). Fujitani teaches this in order to provide a catalyst support with a high compressive strength (col. 8 lines 61-68). Fujitani also teaches the pore diameter to the pore volume distribution in Fig. 2 of this catalyst, and further discloses that the pore diameters of the Fujitani invention are distributed over a very narrow range (col. 7 lines 4-8). It is clear to see from Fig. 2, that 50% - 80% of the total volume comes from the pore with diameters in the range of 180Å – 800Å (see curve 1 in Fig. 2). Fujitani teaches this catalyst and structure this as a successful way of removing NO<sub>x</sub>, CO, and HC from exhaust gasses (see Table 9).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use the catalyst support with the pore diameter and total pore volume to pore size distribution of Fujitani in the exhaust treatment device of Sung in order to successfully remove NO<sub>x</sub>, CO, and HC from exhaust gasses as well as providing a carrier which exhibits a strong compressive strength.

**Regarding claims 3 and 4**, Sung further discloses the preference for greater than 75% of the first catalyst metal and second catalyst metal to be non-alloyed (see col. 8 lines 36-42). As such, having 90 or 95% of the metals non-alloyed would have been obvious to one of ordinary skill in the art at the time of the invention.

**Regarding claims 5 and 6**, Sung further discloses the weight ratio of palladium to rhodium is within the claimed ranges (see col. 20 lines 52-56).

**Regarding claim 9**, Sung further discloses the aluminum oxide comprises gamma aluminum oxide (col. 12 lines 4-9).

4. Claims 12 and 26 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sung et al. (US 5,981,427) and Fujitani et al. (US 4,239,656) as applied to claim 1 above, and further in view of Anatoly et al. (US 6,387,338).

**Regarding claims 12 and 26**, modified Sung discloses the use an oxygen storage component, but fails to teach the exact composition of the claimed oxygen storage component.

Anatoly also discloses oxygen storage materials.

Anatoly teaches an oxygen storage component with the composition of  $\text{Zr}_{0.65}\text{Ce}_{0.25}\text{La}_{0.04}\text{Y}_{0.06}\text{O}_{1.95}$  (see Example 5) in order to enhance the phase stability under high temperature oxidizing and reducing conditions (see Brief Description of Fig. 14).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use the composition of Anatoly in the oxygen storage component of Nunan in order to enhance the phase stability of the oxygen storage component under high temperature oxidizing and reducing conditions which are present in the disclosure of Sung (col. 12 lines 10-19).

5. Claim 13 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sung et al. (US 5,981,427) and Fujitani et al. (US 4,239,656) as applied to claim 1 above, and further in view of Suzuki et al. (US 6,335,305).

**Regarding claim 13**, modified Sung discloses a catalyst for purifying exhaust gasses which contains an oxygen storage component (as discussed above), but fails to teach the oxygen storage component has a stable cubic structure.

Suzuki also discloses a catalyst for purifying exhaust gas (see title).

Suzuki teaches an oxygen storage component with a cubic structure in order to maintain the structure even if a large amount of oxygen is discharged and since oxygen moves freely in the cubic structure, it shows excellent oxygen storage ability as compared to other structures (col. 6 lines 18-24).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use the cubic structure of the oxygen storage component, as in Suzuki, in the oxygen storage component of modified Sung in order to maintain the structure even if a large amount of oxygen is discharged and since oxygen moves freely in the cubic structure, it shows excellent oxygen storage ability as compared to other structures.

6. Claim 25 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sung et al. (US 5,981,427) in view of Fujitani et al. (US 4,239,656) and Foster (US 5,857,140).

**Regarding claim 25**, Sung discloses an exhaust treatment device, comprising:

a substrate;

a 1-catalyst layer (col. 10 lines 45-53) deposited on the substrate (col. 8 lines 25-28), the catalyst layer comprising a first catalyst metal (such as Pd, col. 10 lines 33-43) and a second catalyst metal (such as Rh, col. 10 lines 33-43), wherein greater than or equal to about 70 wt% of the first catalyst metal and the second catalyst metal is non-alloyed under alloying conditions, wherein the weight percent is based on a combined weight of the first catalyst metal and the second catalyst metal (see col. 8 lines 36-43 which

discloses a most preferable embodiment is where greater than 75% of the first and second noble metal components are separate in the later, i.e. will not alloy); and

wherein the first catalyst metal and the second catalyst metal are different and individually selected from the group consisting of palladium and rhodium (col. 10 lines 33-43),

wherein the catalyst layer further comprises an aluminum oxide (col. 7 lines 39-47 and an oxygen storage component (see col. 14 lines 6-12),

wherein the oxygen storage component is represented by the formula  $(\text{Ce}_a\text{Zr}_b\text{La}_c\text{Y}_d\text{Pr}_e\text{O}_x)$ , wherein subscripts a, b, c, d, e, and x, represent atomic fractions, and wherein  $a+b+c+d+e=1$  (see col. 14 lines 6-12, which discloses a composition of oxygen storage material that reads on the claimed composition).

Sung, however, fails to explicitly disclose the aluminum oxide and the storage component have average pore diameters of about 150Å to about 1,000Å, and

Fujitani also discloses a catalyst for purifying exhaust gases and a carrier for the catalyst (see title).

Fujitani, similar to Sung, teaches a catalyst support ( $\gamma\text{-Al}_2\text{O}_3$ , see Example 3 or Example 5 of Fujitani) with an oxygen storage component, also similar to Sung (cerium, col. see Example 3 or Example 5 of Fujitani) and an average pore diameter of 400Å (0.04  $\mu\text{m}$ , see Example 3 or Example 5 of Fujitani). Fujitani teaches this in order to provide a catalyst support with a high compressive strength (col. 8 lines 61-68). Fujitani also teaches the pore diameter to the pore volume distribution in Fig. 2 of this catalyst, and further discloses that the pore diameters of the Fujitani invention are distributed over a



very narrow range (col. 7 lines 4-8). It is clear to see from Fig. 2, that 50% - 80% of the total volume comes from the pore with diameters in the range of  $180\text{\AA} - 800\text{\AA}$  (see curve 1 in Fig. 2). Fujitani teaches this catalyst and structure this as a successful way of removing  $\text{NO}_x$ , CO, and HC from exhaust gasses (see Table 9).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use the catalyst support with the pore diameter and total pore volume to pore size distribution of Fujitani in the exhaust treatment device of Sung in order to successfully remove  $\text{NO}_x$ , CO, and HC from exhaust gasses as well as providing a carrier which exhibits a strong compressive strength.

Sung teaches a catalyst for use in a exhaust treatment device, but fails to teach a retention material disposed around the substrate to form a subassembly and also a housing disposed around the subassembly.

Foster also discloses an exhaust gas treatment device (see Fig. 1)

Foster teaches a retention material (mat, (24)) in order to support the substrate (18) and prevent excessive heat loss (col. 1 line 64 – col. 2 line 5), and also teaches a housing (12) around the substrate and the retention material to improve the durability of the retention material (intumescent material, col. 1 line 64 – col. 2 line 5).

It would have been obvious to one of ordinary skill in the art at the time of the invention to add the retention material and housing of Foster, to the exhaust treatment device of Sung in order to support the substrate and prevent excessive heat loss and to improve the durability of the retention material.

***Response to Arguments***

7. Applicant's arguments filed 8/4/08 have been considered but are moot in view of the new ground(s) of rejection necessitated by amendment.

***Conclusion***

Any inquiry concerning this communication or earlier communications from the examiner should be directed to MATTHEW J. MERKLING whose telephone number is (571)272-9813. The examiner can normally be reached on M-F 8:30-4:30.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Alexa Neckel can be reached on (571) 272-1446. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/M. J. M./  
Examiner, Art Unit 1795

/Alexa D. Neckel/  
Supervisory Patent Examiner, Art Unit 1795